

# Course Specifications

### Valid as from the academic year 2015-2016

## Physics 4: Optics and Physical and Chemical Thermodynamics (O000094)

<b>Course size</b> (nominal values; actual values may depend on programme)				
Credits 5.0	Study time 150 h	Contact hrs	60.0 h	
Course offerings and tea	ching methods in academ	nic year 2016-201	7	
A (semester 2)	guided self-study		10.0 h	
	seminar: coached exe	ercises	20.0 h	
	lecture		20.0 h	
	lecture: plenary exerc	cises	10.0 h	
Lecturers in academic ye	ear 2016-2017			
Zhuiykov, Serge		LA08	lecturer-in-charge	

<b>y</b> , <b>y</b>		5
Offered in the following programmes in 2016-2017	crdts	offering
Bachelor of Science in Food Technology	5	А
Joint Section Bachelor of Science in Environmental Technology, Food Technology and Molecular Biotechnology	5	А
Bachelor of Science in Environmental Technology	5	А
Bachelor of Science in Molecular Biotechnology	5	А

#### **Teaching languages**

English

#### Keywords

Basic Physics, Optics, Thermodynamics, Chemical equilibrium, Molecules and Solids, Processes on solid surfaces

#### Position of the course

The course trains physics, with a focus on both basic principles of optics and thermodynamics and their practical applications. The purpose of the course is to i) make the students familiar with the numerous practical applications of optical devices and their main components as well as with thermodynamic of mixtures, ii) teach students about scientific experiments and measurement methods,

iii) teach students how to report their findings, and

iv) provide the foundations that will allow students to successfully participate in specialize courses.

#### Contents

- 1. Introduction to light: Reflection and refraction
- 2. Lenses and optical instruments
- 3. The wave nature of light; Interference
- 4. Diffraction and polarization
- 5. Thermodynamic aspects of phase transitions
- 6. The thermodynamics of mixtures
- 7. Chemical equilibrium
- 8. Molecules and Solids
- 9. Molecules in motion
- 10. Chemical kinetics
- 11. Processes on solid surfaces

#### **Initial competences**

Competences acquired in Physics 1: Mechanics; Physics 2: Vibration, Waves and Thermodynamics; Physics 3: Electricity and Magnetism

#### **Final competences**

The student will have the ability to describe and analyze both optical and thermodynamic phenomena, to use and apply the various physics laws of optics and thermodynamics. The student will have knowledge about the basic principle of optics, thermodynamic aspects of phase transitions, chemical equilibrium and processes in liquids and on solid surfaces. The student will be able to transfer this obtained knowledge to the modern electrical and optical devices and instruments.

#### **Conditions for credit contract**

Access to this course unit via a credit contract is determined after successful competences assessment

#### **Conditions for exam contract**

This course unit cannot be taken via an exam contract

#### **Teaching methods**

Guided self-study, lecture, lecture: plenary exercises, seminar: coached exercises

#### Learning materials and price

D. C. Giancoli (2009), Physics for scientists & engineers with modern physics, Chapters 32-35, 40, Pearson-Prentice Hall. P. Atkins, J. de Paula, Physical Chemistry, Chapters 4,5,6,19,20,22; Oxford University

P. Atkins, J. de Paula, Physical Chemistry, Chapters 4,5,6,19,20,22; Oxford University Press, 2014.

#### References

#### Course content-related study coaching

#### **Evaluation methods**

end-of-term evaluation and continuous assessment

#### Examination methods in case of periodic evaluation during the first examination period

Written examination with open questions, written examination with multiple choice questions

#### Examination methods in case of periodic evaluation during the second examination period

#### Examination methods in case of permanent evaluation

Participation

#### Possibilities of retake in case of permanent evaluation

examination during the second examination period is possible in modified form

#### Extra information on the examination methods

End-of-term evaluation and continuous assessment

#### Calculation of the examination mark

Final written exam with open questions and with multiple choice questions: 80% Seminar Participation: 20%